

FACULTY OF CHEMISTRY					
SUBJECT CARD					
Name of subject in English:	Nanostructures in Industrial and Numerical Applications				
Main field of study (if applicable):	Chemical and Process Engineering				
Specialization (if applicable):	Chemical Nano-engineering				
Profile:	academic				
Level and form of studies:	2nd level, full-time				
Kind of subject:	obligatory				
Subject code:	ICC025007				
Group of courses:	YES				
	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	30	30			30
Number of hours of total student workload (CNPS)	60	60			30
Form of crediting	crediting with grade	crediting with grade			crediting with grade
For group of courses mark (X) final course					
Number of ECTS points	2	2			1
including number of ECTS points for practical (P) classes		2			1
including number of ECTS points for direct teacher-student contact (BK) classes	1	1			1
PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES					
1. Knowledge of the basic quantum mechanics					
SUBJECT OBJECTIVES					
C1 Gaining basic knowledge of computational quantum chemistry					
C2 Gaining knowledge how to select the proper computational technique for application in nano-engineering					
C3 Application of computational quantum chemistry methods to model properties of light-driven molecular motors					
SUBJECT EDUCATIONAL EFFECTS					
related to knowledge:					
PEK_W01 Student has knowledge regarding the variety of quantum-chemistry methods					
PEK_W02 Student can choose the right computational approach to specific properties of nano-systems					
PEK_W03 Student knows the basic principles of operation of light-driven molecular machines					
related to skills:					
PEK_U01 Student can work in the high-performance computing center environment					
PEK_U02 Student can run efficiently quantum-chemistry programs					
PEK_U03 Student can analyze the results of quantum-chemistry calculations					
PEK_U04 Student can pinpoint the operation mechanism of light-driven molecular machines using computational quantum chemistry methods					
related to social competences:					

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PEK_K01	student understands the need to inform the public about the need to achieve the goals of development of nanoengineering	
PEK_K02	student is able to work in a group, performing various roles including group leader	
PEK_K03	student is aware of the social role of the engineer	
PEK_K04	student is ready to critically evaluate his/her knowledge and received content	
PROGRAMME CONTENT		
Lectures		Number of hours
Lec 1	Applications of computational quantum chemistry in modeling of nanostructures – an overview	2
Lec 2	The basics of molecular quantum mechanics	2
Lec 3	Hartree-Fock self-consistent-field method	2
Lec 4	Density functional theory. Kohn-Sham method.	2
Lec 5	Time-dependent density functional theory: formalism and applications.	2
Lec 6	Gaussian basis sets for molecular calculations	2
Lec 7	Methods to account for environmental effects	2
Lec 8	Electron correlation: Møller-Plesset perturbation theory	2
Lec 9	Electron correlation: Coupled-cluster theory	2
Lec 10	Electron correlation: Multiconfigurational self-consistent-field method	2
Lec 11	Applications: Modeling of excited states	2
Lec 12	Applications: Light-driven molecular motors (part 1)	2
Lec 13	Applications: Light-driven molecular motors (part 2)	2
Lec 14	Applications: Light-driven molecular motors (part 3)	2
Lec 15	Exam	2
	Total hours	30
Classes		Number of hours
Class 1	Introduction to high-performance computer center environment	2
Class 2	Quantum chemistry computer tools: an introduction.	2
Class 3	Building structures, geometry optimization, vibrational analysis	2
Class 4	Building structures, geometry optimization, vibrational analysis (cont.)	2
Class 5	Hartree-Fock self-consistent-field method	2
Class 6	Density functional theory and its time-dependent extension	2
Class 7	Methods to account for environmental effects	2
Class 8	Møller-Plesset perturbation theory and coupled-cluster theory	2
Class 9	Project 1: Modeling of spectra of molecules in solution (individual assignments)	2
Class 10	Project 1: Modeling of spectra of molecules in solution (individual assignments)	2
Class 11	Multiconfigurational self-consistent field method	2
Class 12	Project 2: Light-driven molecular motors (individual assignments)	2
Class 13	Project 2: Light-driven molecular motors (individual assignments)	2
Class 14	Project 2: Light-driven molecular motors (individual assignments)	2

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Class 15	Project 2: Light-driven molecular motors (individual assignments)	2
	Total hours	30
Seminar		Number of hours
Sem 1	Introductory classes	2
Sem 2	Individual assignments	2
Sem 3	Application of multiscale modeling in nanotechnology	2
Sem 4	Fundamentals of molecular dynamic simulations	2
Sem 5	Thermostats and barostats	2
Sem 6	Force-fields	2
Sem 7	Modeling in transport using molecular dynamics	2
Sem 8	Predicting properties using molecular dynamics	2
Sem 9	Combining molecular dynamics with other modeling techniques	2
Sem 10	Individual assignments (assessment part 1)	2
Sem 11	Project 1: presentations	2
Sem 12	Molecular machines and molecular dynamics (part 1)	2
Sem 13	Individual assignments (assessment part 2)	2
Sem 14	Molecular machines and molecular dynamics (part 2)	2
Sem 15	Project 2: presentations	2
	Total hours	30
TEACHING TOOLS USED		
N1. Lecture with multimedia presentation N2. Hands-on sessions using computers N3. Preparation of reports		
EVALUATION OF SUBJECT EDUCATIONAL EFFECTS ACHIEVEMENT		
Evaluation (F – forming (during semester), P – concluding (at semester end))	Educational effect number	Way of evaluating educational effect achievement
P (lecture)	PEK_W01 - PEK_W03	Written evaluation
F1-F15 (class)	PEK_U01 - PEK_U04	Evaluation of individual reports
F1-F15 (seminar)	PEK_U03 – PEK_U04 PEK_K01 – PEK_K04	Evaluation of the student's project presentation
PRIMARY AND SECONDARY LITERATURE		
<u>PRIMARY LITERATURE:</u>		
[1] Ira N. Levine, „Quantum Chemistry”, 7th Edition, Pearson Education, 2014.		
<u>SECONDARY LITERATURE:</u>		
[1] B. Roos, R. Lindh, P. A. Malmqvist, V. Veryazov, P. O. Widmark, „Multiconfigurational Quantum Chemistry”, 1st Edition, Wiley, 2016.		
[2] W. Koch, M. C. Holthausen, „A Chemist's Guide to Density Functional Theory”, 2nd Edition, Wiley, 2000.		
SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)		
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